

Traditio et Innovatio

Waste water recycling products as Phosphor (P) fertilizer Selected findings of one pot experiment

Telse Vogel, Michael Nelles, Bettina Eichler-Löbermann

Introduction / Objectives

The efficient use of P in agriculture as well as P recycling from waste products for food production is on primary concern. Thereby, recycling P from waste water residuals is of particular importance due to their high P concentration, as about 3 MT P globally incurred each year.

The objectives of this study were to determine how (i) different P recycling products from waste water residuals affect the P uptake of rye and amaranth and (ii) how the recycling products affect the P pools in soil in dependency of their elemental composition, which differed according to the used waste water residual and used recycling process.

Method

In the pot experiment the P fertilizing effect of

- 1. one untreated sewage sludge ash (ut-SSA) based on AI precipitation \rightarrow AI(ut)-SSA
- 2. one sulfuric acid digested sewage sludge ash (SSA) based on AI precipitation \rightarrow AI-SSA
- 3. one thermo-chemical treated SSA prepared with $CaCl_2 \rightarrow Ca-SSA$
- 4. one thermo-chemical treated SSA prepared with MgCl₂ \rightarrow Mg-SSA
- 5. magnesium ammonium phosphate recovered from digested sludge by precipitation \rightarrow struvite

were analyzed in comparison to one phosphate rock based fertilizer (Triple superphosphate, **TSP**) and one control treatment without P application (**CON**) (Tab.1).

Pot experiment: Mitcherlich pots were filled with 6 kg of a P poor loamy sand and were amended with the treatments at a rate equivalent to 200 mg P per pot. After 8 week growing time plant and soil samples were analyzed according to P uptake from the crops and the soil P pools.

Table 1 Total P (Pt) content and solubility of P (Citric acid soluble P, Pca) as well as total nutrient concentration (K, Mg, Ca, N, Al, Fe) of the fertilizer treatments

	P _t	P_{ca}	K	Mg	Ca	Ν	Al	Fe
g kg⁻¹								
AI-(ut)SSA	98.6	29.9	9.5	15.1	76.1	n.d.	95.6	20.6
AI-SSA	58.5	45.9	5.1	7.7	42.6	n.d.	55.8	10.4
Ca-SSA	62.4	46.0	4.1	9.3	96.9	n.d.	23.9	29.2
Mg-SSA	65.9	44.0	4.8	49.1	54.3	n.d.	17.1	38.1
Struvite	121.7	102.7	0.6	93.4	4.9	41.7	39.6	18.5
TSP	157.0	168.9	n.d.	8.9	106.7	n.d.	29.4	22.5

Results

- Among all fertilizer treatments struvite increased the P uptake and the plant available P pools (Pdl, NaHCO3) fraction) in soil the most compared to CON followed by the Mg-SSA (Fig. 1 and 2 (1.))
- 2. The Ca-SSA had a lower P fertilizing effect compared to the Mg-P based recycling products due to its high concentrations in low soluble P compounds, which was visible in the highest increase of the H₂SO₄ fraction (P associated with Ca and Mg minerals) in soil compared to the other fertilizer treatments (Fig. 2 (3.)).
- 3. The AI-SSA resulted also in a lower P fertilizing effect compared to the Mg-P based products, because of its high concentrations in amorphous AI, which may increased the P sorption capacity in soil and increase the risk of Al-toxicity for crops. This is confirmed by the highest increase of the NaOH fraction (P sorbed and fixed by Al and Fe oxides) in soil after application of the AI-SSA compared to the other fertilizer treatments (Fig. 2 (2.)).

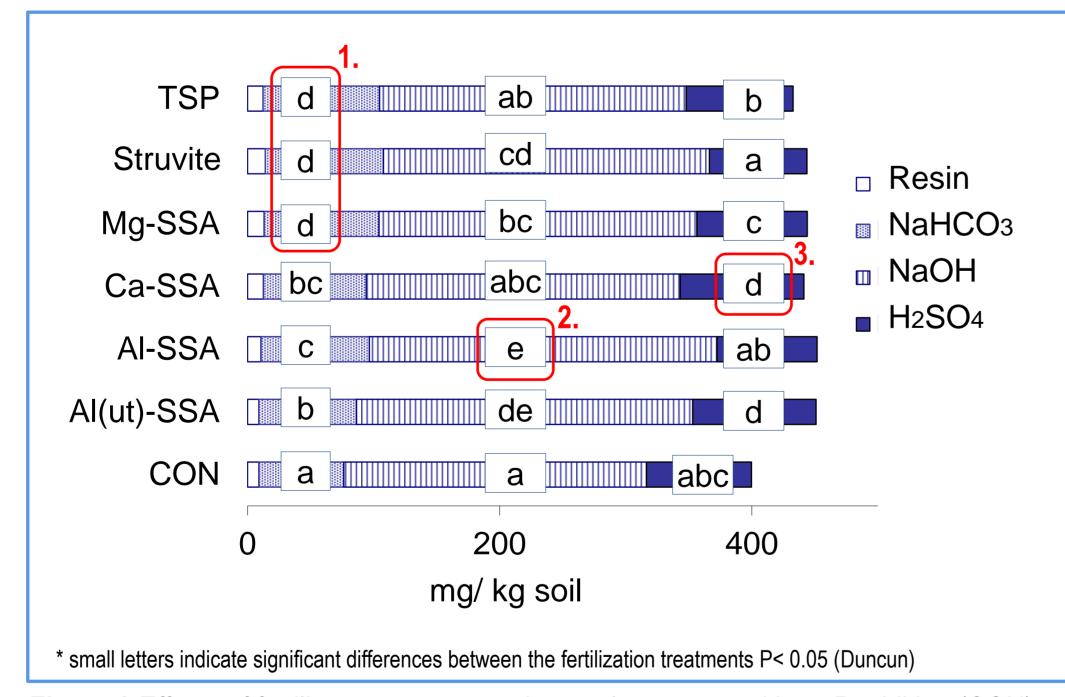


Figure 2 Effects of fertilizer treatments and control treatment without P addition (CON) on proportions of sequentially extracted P fraction in soil in average after cultivation of amaranth and rye

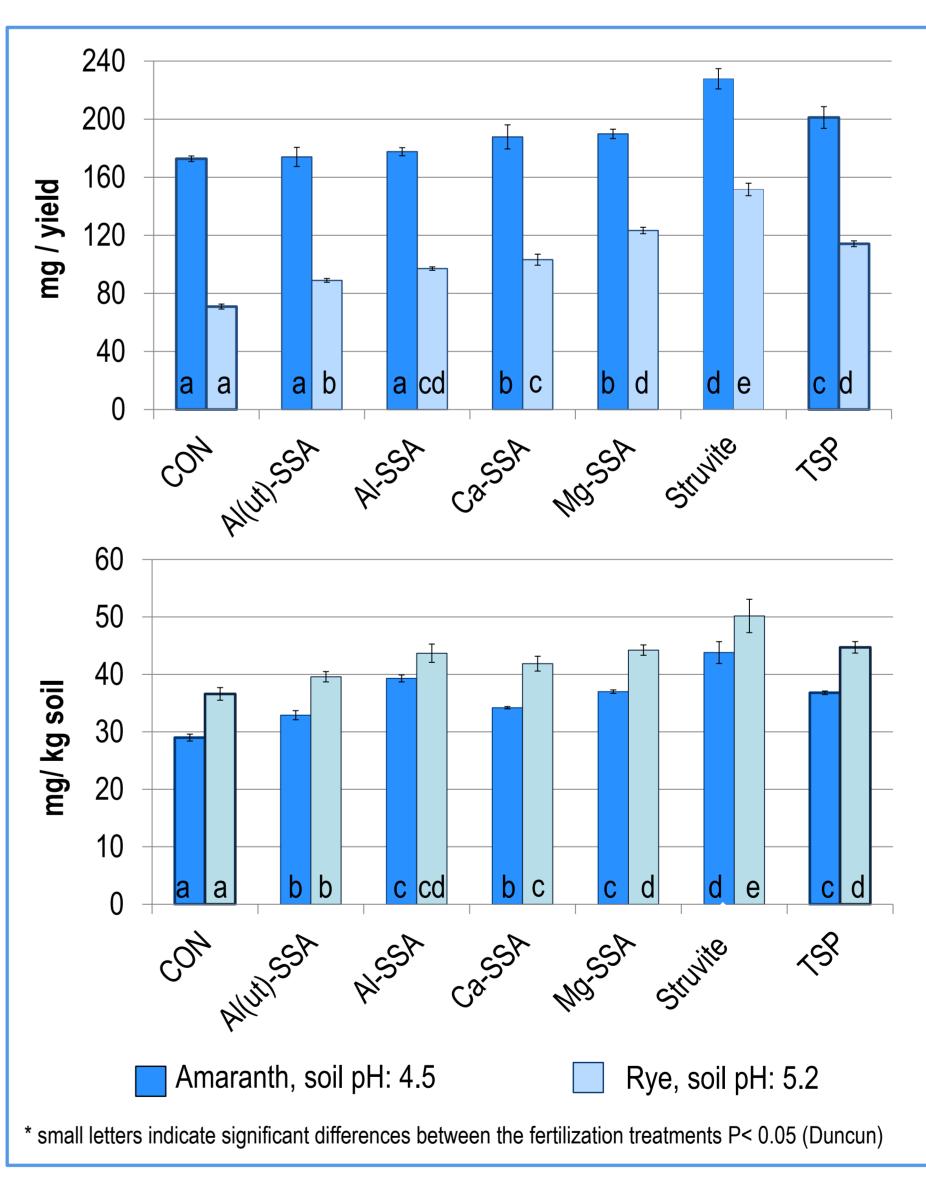


Figure 1 P uptake by amaranth and rye (upper figure) as well as doublelactate

soluble P (Pdl) concentration in soil (figure below) after cultivation of amaranth and rye and in control treatment and after addition of fertilizer treatments

Conclusion

- 1. The Mg-P based recycling products, struvite and the Mg-SSA, are effective sources for P fertilization on acidic loamy sand.
- 2. For thermo-chemical treatment the use of a Mg donor (as MgCl₂) should be preferred, because of the higher P fertilizing effect of the Mg-SSA compared to Ca-SSA.
- 3. The AI-SSA should not be applied under acidic soil conditions (pH < 5) due to its high concentration in amorphous AI, which may increase the sorption capacity in soil or may even has a toxic effect on crops.
- 4. The Al(ut)-SSA is not suitable as P fertilizer according to its low P availability for crops and its high concentration in heavy metals.

University of Rostock | Faculty of Agriculture and Environmental Sciences | Justus-von-Liebig Weg 6 18051 Rostock, German Telse Vogel, Michael Nelles, Bettina Eichler-Löbermann telse.vogel@uni-rostock.de